

# MXenes as Photocatalytic Materials for Water Splitting

Global Warming  
↓  
Environmental Pollution

Need for **green** fuels

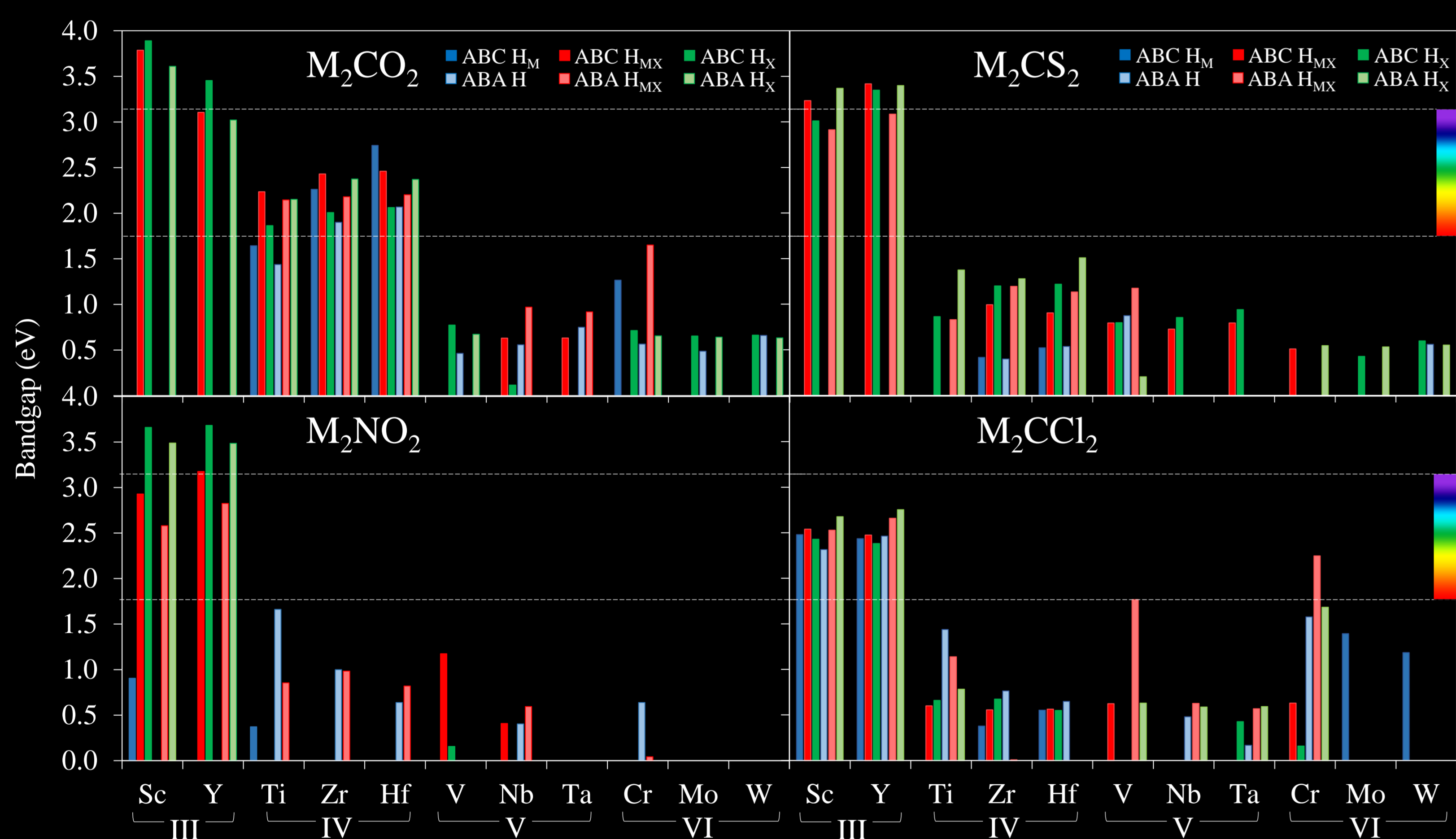
Hydrogen (H<sub>2</sub>)



## OBJECTIVES

Modify and design the bandgap of MXenes by varying their composition (M, X, T), width (n), stacking and termination position, to find potential photoactive candidates for water splitting using solar light.

## Density of States



- The band alignment with respect to the half-reaction potentials has been studied for the most promising photoactive cases ( $E_g > 1.23$  eV).
- The ideal cases will be those that, in addition to having a suitable band alignment, are the most stable structure among the six considered.
- Several structures from Groups III and IV exhibit correct alignments.
- The cases of  $Zr_2CO_2$ ,  $Sc_2CCl_2$ ,  $Y_2CCl_2$ ,  $Sc_2CS_2$ , and  $Y_2CS_2$  fulfill these optimal conditions, which allows us to propose them as potential candidates for the photocatalytic water splitting.
- These systems exhibit an indirect bandgap and a significant charge density separation between the valence band maximum and conduction band minimum (VBM and CBM), promoting the separation of generated charges.

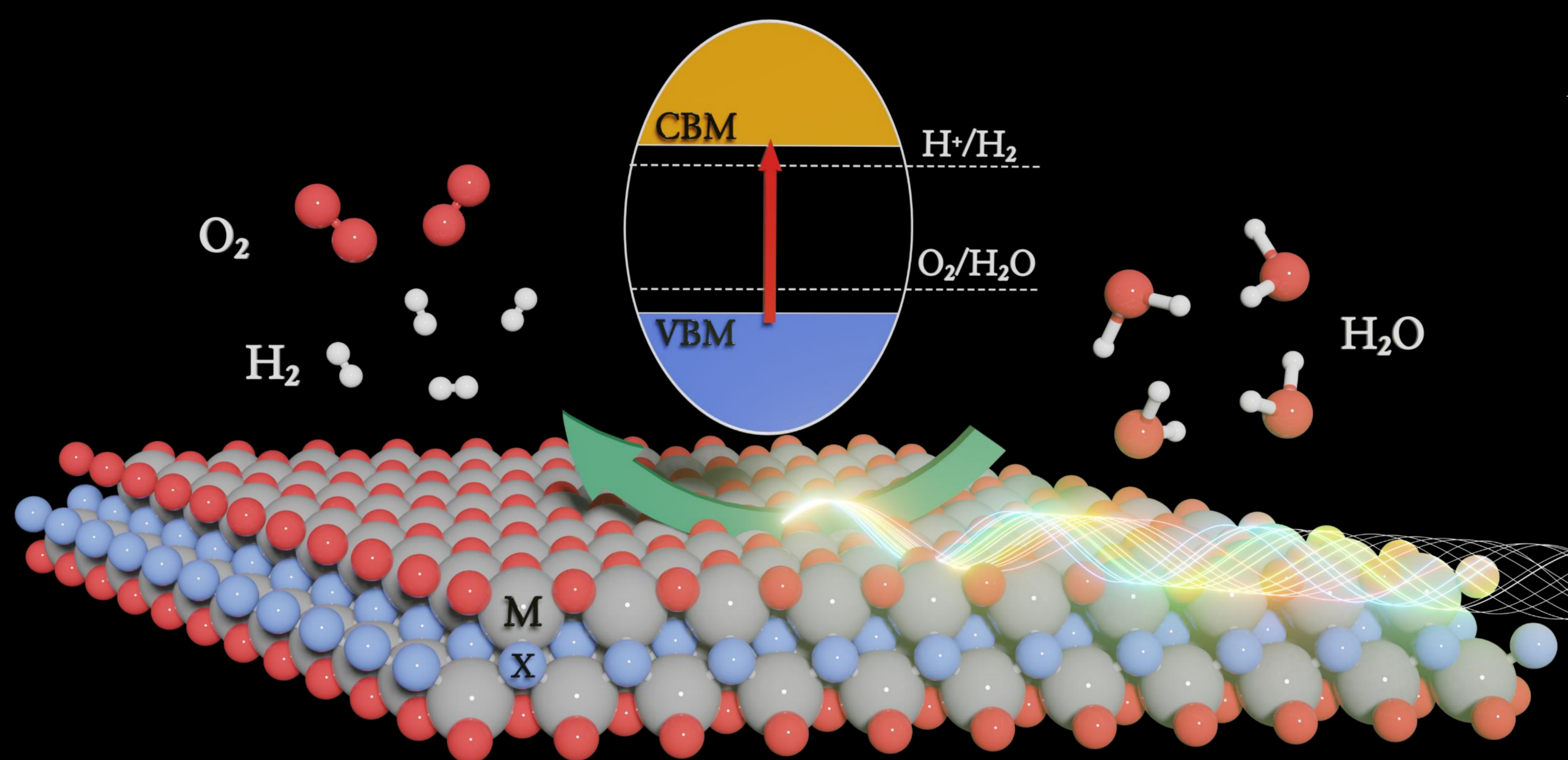
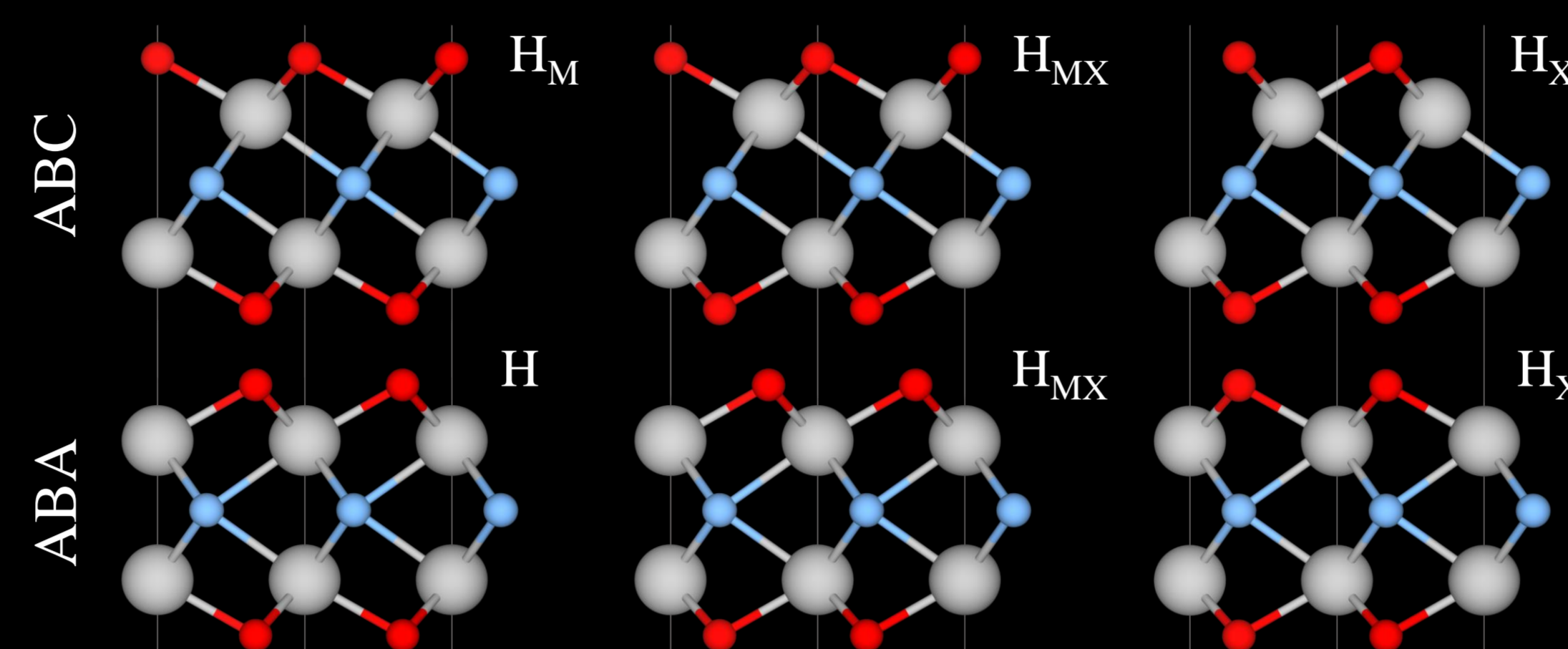
## CONCLUSIONS

Based on DFT calculations, MXenes with  $n = 1$ ,  $X = C$ ,  $M =$  Groups III and IV, and  $T = O, S, Cl$ , especially  $Zr_2CO_2$ ,  $Sc_2CCl_2$ ,  $Y_2CCl_2$ ,  $Sc_2CS_2$ , and  $Y_2CS_2$ , exhibit a visible range bandgap, optimal for solar light absorption, and band edges that exceed the half-reaction potentials of water splitting, suitable for photocatalysing the process and generating green H<sub>2</sub>.

**MXenes**<sup>[1]</sup>  
(M<sub>n+1</sub>X<sub>n</sub>T<sub>x</sub>)

M = Transition Metal (Groups III – VI)  
X = C or N n = 1–3  
T = Termination (p-block: O, F, OH, H, S, Cl)

Semiconductors when adding a termination.<sup>[2]</sup>  
Good candidates for photocatalysing the water splitting process and produce clean H<sub>2</sub>.<sup>[3]</sup>



Sustainable source of H<sub>2</sub>:  
**Water Splitting**  
 $2 H_2O \rightarrow 2 H_2 + O_2$

**Problem:**  
Requires large amounts of energy

**Solution:**  
Sunlight as energy source

**Photocatalysis**



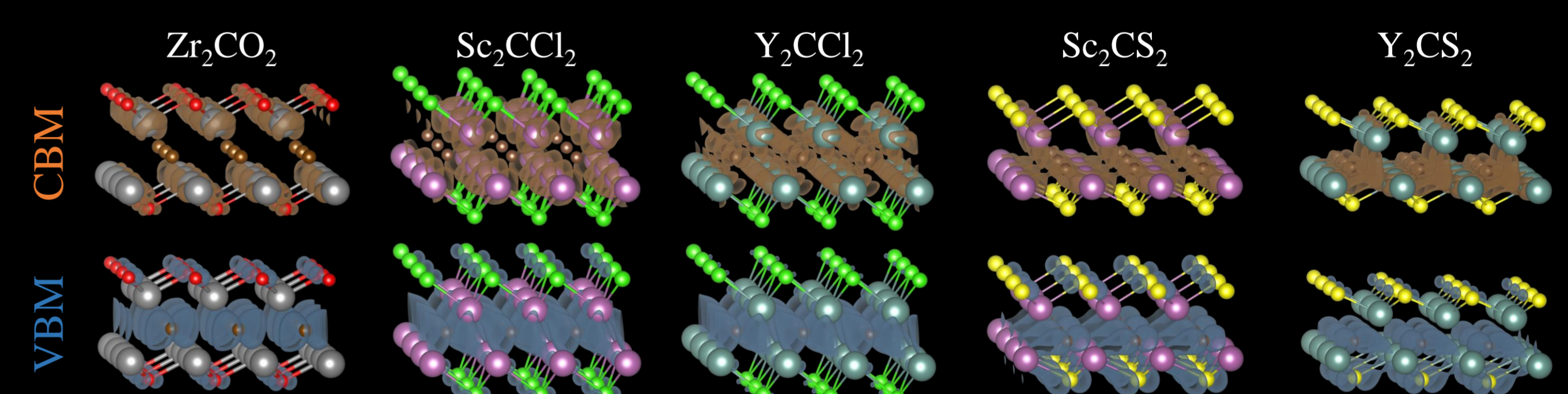
## TOOLS

**Computational Method:** DFT

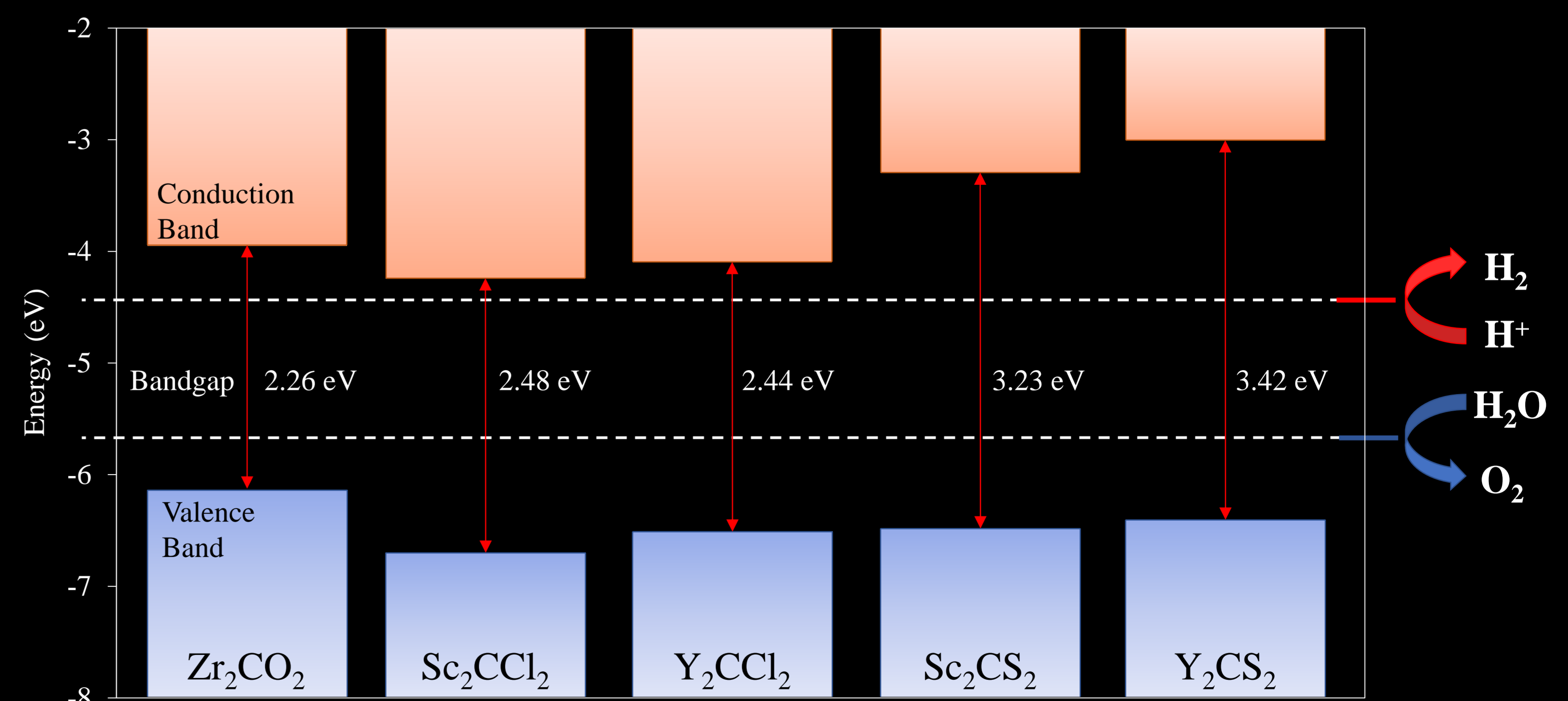
**Functionals:** PBE and PBE0

**Structures:** periodic slab models, considering two stackings (ABC and ABA) and three termination positions for each stacking (H<sub>M</sub>/H, H<sub>MX</sub> and H<sub>X</sub>)

- 6 different structures for each terminated MXene are considered (2376 in total).
- Group III and IV MXenes with  $n = 1 \rightarrow$  large bandgaps, in the visible region.  
→ The most promising cases for being photoactive materials with sunlight.
- C-MXenes  $\rightarrow$  more semiconducting cases and larger bandgaps than N-MXenes.
- Pristine MXenes  $\rightarrow$  metallic properties (not photoactive).
- MXenes  $n = 2, 3 \rightarrow$  Increasing the amount of "bulk" tends to make them metallic.



## Band Alignment



## REFERENCES

- [1] *Adv. Mater.* 2011, 23, 4248–4253.  
[2] *Adv. Funct. Mater.* 2013, 23, 2185–2192.  
[3] *J. Mater. Chem. A* 2016, 4, 11446–11452.

